TABLE OF INCOMPATIBLES.

Substances	*Incompatible with*
Acidum Citricum	Alkalies. Alkaline Solutions. ——— Sulphates. ——— Carbonates. ——— Tartrates. Soaps. Earthy Carbonates. Acetates. Metallic Carbonates.
—— Muriaticum	Alkalies. Tartrate of Potass. Sulphuret of Potass. Most earths. —— Oxides. —— Carbonates of the incompatible oxides. Tartarized Antimony. Tartrate of Iron. Nitrate of Silver. Solution of Subacetate of Lead.
—— Nitricum	Alkalies. Carbonates of Alkalies. Acetates of Alkalies. Earths. Metals, except Platina and Gold. Oxides. Sulphate of Iron. Solution of Acetate of Lead. Sulphurets. Charcoal. Phosphorus. Sugar. Alcohol and Spirits.
—— Sulphuricum	Alkalies. Alkaline Carbonates. Some Earths. —— Earthy Carbonates. Solution of Muriate of Lime. Barytic Salts. Most Metals. —— Oxides of Metals. Solution of Acetate of Lead.
—— Tartaricum	Alkalies. Carbonates of Alkalies. Salts of Potass. Most Earths. —— Carbonates of Earths. Salts of Lime. ——— Lead.
Ammoniæ Liquor	All Acids. Saline solutions of most Earths, except Barytes or Lime. ——— ——— ——— Metals, except Barytes or Lime.

Substances	*Incompatible with*
Ammoniæ Subcarbonas -	Acids. Potash. Soda. Subcarbonate of Soda. ——— Potash. Supersulphate of Potass. Supertartrate of Potass. Lime. —— Water. Solution of Muriate of Lime. Magnesia. Sulphate of Magnesia. Alum. Solutions of Iron, except of Tartarized Iron. Sulphate of Iron. ——— Zinc. Oxymuriate of Mercury. Acetate of Lead. Subacetate of Lead.
——— Acetatis Liquor -	Acids. Potass. Soda. Subcarbonate of Potass. ——— Soda. Lime Water. Magnesia. Sulphate of Magnesia. Oxymuriate of Mercury. Sulphate of Iron. —— —— Copper. ——— Zinc. Nitrate of Silver. Acetate of Lead. Subacetate of Lead.
Potassæ Subcarbonas -	Acids. Acidulous Salts. Muriate of Ammonia. Acetate of Ammonia. Lime Water. Muriate of Lime. Sulphate of Magnesia. Alum. Tartarized Antimony. Nitrate of Silver. Ammoniated Copper. ——— Iron. Tincture of Ammoniated Iron. ——— Muriated Iron. Sulphate of Zinc. Calomel. Corrosive Sublimate. Acetate of Lead. Subacetate of Lead.
——— Subcarbonatis Liquor	The same as with Potassæ Subcarb.
——— Carbonas - -	Nearly the same as with Potassæ Subcarb. except Sulphate of Magnesia. Calomel, unless heated.
——— Liquor - - -	Acids. Acidulous Salts. Subcarbonate of Ammonia. Acetate of Ammonia

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Substances	*Incompatible with*
Potassæ Liquor	Muriate of Ammonia. Preparations of Earths } held in solution by ———— Metals } Acids. Calomel. Corrosive Sublimate.
—— Acetas	Sulphuric Acid. Muriatic ——. Nitric ——. Sulphate of Soda. ———— Magnesia. Most Metallic Salts. —— Earthy Salts.
—— Tartras	—— Acids. —— Acidulous Salts. Lime Water. Muriate of Lime. Salts of Lead. —— Silver.
—— Supersulphas	Alkalies. Earths. Carbonates of Earths. Most Metals. —— Oxides.
—— Sulphas	Tartaric Acid. Barytes Water. Muriate of Barytes. ———— Lime. Acetate of Lead. Subacetate of Lead.
Sodæ Subcarbonas	Acids. Acidulous Salts. Muriate of Ammonia. Earthy and Metallic Salts. Lime Water.
—— Carbonas	The same as those of Sodæ Subcarbonas.
—— Sulphas	Subcarbonate of Potash. Solution of Barytes. Barytic Salts. Muriate of Lime. Nitrate of Silver. Acetate of Lead. Subacetate of Lead.
—— Tartarizata	Most Acids. —— acidulous salts except Supertartrate of Potash. Barytic Salts. Salts of Lime. Acetate of Lead. Subacetate of Lead.
Alumen exsiccatum	Alkalies. Carbonates of Alkalies. Tartrate of Potash. Lime. Lime Water. Magnesia. Carbonate of Magnesia. Acetate of Lead.
Creta Preparata	Acids. Acidulous Salts.
Calx	Acids. Acidulous Salts. Alkaline Carbonates.

Substances	*Incompatible with*
Calx	Ammoniacal Salts. Borates. Metallic Salts. Astringent Vegetable Infusions.
Calcis Liquor	The same substances as with Calx.
—— Murias	Sulphuric Acid. Sulphates. Potash. Soda. Carbonate of Potash. ———— Soda. ———— Ammonia.
—— Muriatis Liquor	Same as with Calcis Murias.
Magnesiæ Sulphas	Potash. Soda. Ammonia. Subcarbonate of Potash. ———— Soda. Lime Water. Muriate of Lime. Acetate of Lead.
———— Subcarbonas	Acids. Acidulous Salts. Muriate of Ammonia. Lime Water. Metallic Salts.
Magnesia	Same as with Magnesiæ Subcarb. except Lime Water.
Antimonium Tartarizatum	Its solution is incompatible with Alkalies. Alkaline Carbonates. Some Earths. Lime Water. Muriate of Lime. Some Metals. —— Oxides of Metals. Acetates of Lead. Infusions of Cinchona. ———— Rhubarb. ———— Catechu.
Argenti Nitras	Almost all Spring Water. ———— River Water. Potash. Soda. Carbonate of Potash. ———— Soda. Soaps. Lime Water. Sulphuric Acid, Muriatic ——, Tartaric ——, and Salts containing these Acids. Carbonate of Ammonia. Liquor Arsenicalis. Sulphuretted Hydrogen. Hydrosulphurets. Astringent Vegetable Infusions.
Liquor Arsenicalis	Acids. Acidulous Salts. Lime Water. Muriate of Lime. Sulphate of Magnesia. Alum.

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Substances	*Incompatible with*
Liquor Arsenicalis	Sulphate of Iron. Muriate of Iron. Nitrate of Silver. Sulphate of Copper. Sulphuretted Hydrogen and its compounds. Decoction of Cinchona.
Cuprum Ammoniatum	Acids. Potash. Soda. Lime Water.
Ferri Sulphas	Potash. Soda. Ammonia. Carbonate of Potash. ——— Soda. Barytes. Salts of Barytes. ——— Strontia. Strontia. Lime Water. Muriate of Lime. Soaps. Nitrate of Silver. Acetates of Lead. Astringent Vegetable bodies.
—— Subcarbonas	Acids. Acidulous Salts.
—— Muriatis tinctura	Alkalies. Carbonates of Alkalies. Lime Water. Carbonate of Lime. Magnesia. Carbonate of Magnesia. Astringent Vegetable bodies. Solution of Gum Arabic.
Ferrum Ammoniatum	Alkalies. Alkaline Carbonates. Lime Water. Astringent Vegetable bodies.
Ferri Ammoniati Tinctura	The same as with Ferrum Ammoniatum.
—— Vinum	The same as with Ferri Muriatis Tinctura.
Hydrargyrum cum Cretâ	Acids. Acidulous Salts.
Hydrargyri oxydum Rubrum	Acids. Acidulous Salts. Sulphuretted Hydrogen.
——— Nitrico Oxydum	The same as with Hydrargyri Oxydum Rubrum.
——— Oxymurias	Potash. Soda. Ammonia. Carbonate of Potash. ——— Soda. ——— Ammonia. Sulphuret of Potash. All Hydrosulphurets. Soap. Lime Water. Many Metals. Tartarized Antimony. Nitrate of Silver. Acetates of Lead.

Substances	*Incompatible with*
Hydrargyri Oxymurias -	Infusion of Bitter Vegetables. ——— Astringent Vegetables.
——— Submurias -	Potash. Soda. Ammonia. Alkaline Carbonates. Hydrosulphurets. Lime Water. Iron. Lead. Copper.
Plumbi Acetas - -	Sulphuric Acid. Muriatic ——. Carbonic ——. Citric ——. Tartaric ——. Potash. Soda. Ammonia. Muriate of Soda. Liquor Ammoniæ Acetatis. Lime Water. Carbonate of Lime. Sulphate ———. Solution of Sulphuretted Hydrogen. Hard Water usually.
——— Subacetatis Liquor -	Same as with Plumbi Acetas.
Zinci Sulphas - -	Alkalies. Alkaline Carbonates. Lime Water. Hydrosulphurets. Astringent Vegetable Infusions.
—— Oxydum - - -	Acids. Acidulous Salts. Alkalies.
Potassæ Sulphuretum -	Acids which combine with Potash and expel Sulphuretted Hydrogen Gas. Solutions of most of the Metals.
nfusum Anthemidis - -	Salts of Iron. ——— Mercury. ——— Silver. ——— Lead.
——— Armoraciæ Comp.	Alkaline Carbonates. Salts of Silver. ——— Mercury.
——— Calumbæ - -	Lime Water. Acetates of Lead. Oxymuriate of Mercury.
——— Caryophyllorum	Lime Water. Solutions of the Preparations of Iron. ——————— Zinc. ——————— Lead. ——————— Silver. ——————— Antimony.
——— Cascarillæ - ——— Cuspariæ -	The same as with Infusum Caryophyllorum. Solutions of the Salts of most Metals.
——— Digitalis -	——————— Iron. Probably by those of most other Metals.
——— Gentianæ Comp.	Solution of Acetate of Lead. ——— Sulphate of Iron.
——— Lini Compositum -	Preparations of Lead. ——————— Iron. Most Metallic Salts.

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Substances	*Incompatible with*
Infusum Rhei - -	The stronger Acids. Metallic Solutions. Some astringent infusions.
——— Rosæ Compositum	Alkalies. Earths and all substances which combine with Sulphuric Acid, or are acted upon by small quantities of it. Acetate of Lead. Sulphate of Iron.
——— Sennæ Compositum	Strong Acids. Lime Water. Most Metallic Salts.
Simarubæ - -	Alkaline Carbonates. Lime Water. Salts of Lead. ——— Silver. ——— Mercury.
Decoctum Aloes Compositum	Acids. Acidulous Salts. Earthy Salts. Metallic Salts. All substances which are decomposed by, or which decompose Subcarbonate of Potash.
——— Cydoniæ - -	Acids. Alcohol. Most Metallic Solutions.
——— Quercûs - -	Alkaline Solutions. Most Metallic Salts. Solutions of Isinglass. Decoction of Yellow Bark.
——— Sarsaparillæ -	Lime Water. Acetate of Lead.
Mistura Ferri Compositum	Acids, Acidulous Salts } which dissolve the Proto-carbonate of Iron. Vegetable astringents.
Spiritus Ammoniæ Aromaticus	Acids. Acidulous Salts. Earthy ——. Lime Water. Metallic Salts.
——— Fœtidus	The same as in Sp. Amm. Arom.
——— Succinatus	Acids. Acidulous Salts. Earthy ——. Metallic ——.
——— Camphoræ	Water.
Tinctura Opii - - -	Potash and its Subcarbonate. Soda ——. Ammonia ——. Most Metallic Salts. Infusion of Galls.